

Block Copolymers Containing a Hydrophobic Domain of Membrane-Lytic Peptides Form Micellar Structures and Are Effective Gene Delivery Agents

Joan G. Schellinger, Joshuel A. Pahang, Julie Shi, and Suzie H. Pun*

Department of Bioengineering and Molecular Engineering & Sciences Institute, University of Washington, 3720 15th Avenue NE, Box 355061, Seattle, Washington 98195, United States

Supporting Information

ABSTRACT: Endosomal release peptides have been incorporated in synthetic gene delivery formulations to increase transfection efficiencies. In this work, cationic copolymers containing sHGP, a membrane-lytic peptide derived from HIV gp41, were synthesized and evaluated. Diblock, with sHGP displayed on one block, and statistical, with sHGP randomly displayed, copolymers were prepared via reversible addition—



fragmentation chain transfer (RAFT) polymerization. While the statistical copolymers existed as unimers in solution, amphiphilic diblock copolymers self-assembled into cationic micelles in aqueous solution as evidenced by transmission electron microscopy (TEM) and dynamic light scattering analyses. This self-assembly sequestered the lytic domain and significantly reduced the cytotoxicity of the materials. However, when complexed with plasmid DNA, both the diblock and statistical copolymers of sHGP showed higher gene delivery efficacy compared to the copolymers without the membrane lytic motif. The ability of amphiphilic, diblock copolymers containing endosomal release motifs to self-assemble and sequester lytic domains is a promising feature for the nucleic acid delivery.

E ndosomal release and cytosolic delivery of therapeutics still remain a major barrier in achieving an efficient nonviral nucleic acid therapy. Macromolecular cargos that enter cells by endocytosis will become inactivated in lysosomal compartments unless released from this degradatory pathway.¹ Thus, recent developments in nucleic acid delivery include the development of multifunctional vehicles that can overcome barriers such as endosomal escape. These biomaterials enable productive delivery of genes and oligonucleotide therapeutics to mammalian cells.^{2,3}

Peptides are a promising class of bioactive molecules that can be easily functionalized and incorporated into polymeric systems. Membrane-active peptides such as melittin or INF7 have been adapted as endosomal release peptides in intracellular delivery formulations.⁴ The use of these and other peptides in gene delivery vehicles was reviewed recently.⁵ These bioactive peptides are incorporated for nucleic acid complexation, as targeting ligands, and to overcome entrapment from the endocytic pathway. Peptides can therefore function as key components of delivery vectors by facilitating efficient gene expression. Previously, we reported a 15-amino acid peptide, sHGP, derived from the endodomain of HIV gp41, that displays membrane-lytic ability.^{6,7} Conjugation of sHGP to branched polyethylenimine (PEI) led to enhanced transfection efficiency compared to unmodified PEI and scrambled sHGPmodified PEI.⁸ However, disadvantages of using PEI-based materials in vivo include their lack of degradability, their propensity to induce erythrocyte aggregation, and their ability to activate the complement cascade.^{9,10}

Herein, we report the incorporation of sHGP into a peptidebased polymeric system composed of oligopeptide monomers (K₁₀) (for nucleic acid binding) and *N*-(2-hydroxypropyl)methacrylamide (HPMA) in two different architectures.¹¹ We utilize a modular synthesis involving reversible addition fragmentation chain transfer (RAFT) polymerization that allows us to incorporate different bioactive peptides, such as endosomal escape peptides, in a block¹² or statistical¹³ form. Polymerization via the RAFT method offers advantages in preparation of polymers with narrowly dispersed molecular weights and well-defined composition and structure for gene transfer.^{14,15}

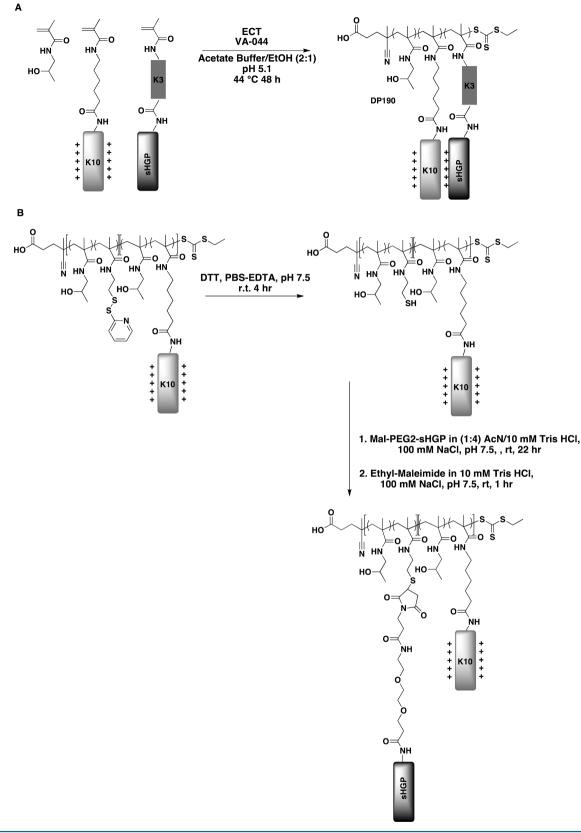
Statistical copolymers containing sHGP (st-sHGP) were prepared by copolymerization of HPMA, methacrylated-*oligo*-Llysine (MaAhxK₁₀), and methacrylated-K₃-sHGP (MaK₃sHGP) using a trithiocarbonate-based chain transfer agent in a 2:1 acetate buffer/ethanol solvent mixture (Scheme 1A). The monomer feed ratio and the degree of polymerization were chosen based on our previous optimization studies: 77% HPMA, 20% MaAhxK₁₀, 3% MaK₃sHGP at degree of polymerization (DP) of 190. Three lysine residues were included as a linker in the methacrylamido sHGP to improve its solubility for polymerization. To synthesize the noncleavable sHGP diblock copolymers (*b*-sHGP), we took the previously reported diblock poly((HPMA-*co*-PDSMA)(HPMA-*co*-

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Scheme 1. (A) Synthesis of Statistical sHGP Copolymers and (B) Synthesis of sHGP-Grafted Diblock Copolymers

 $MaAhxK_{10}$) and reduced the disulfide moieties using dithiothreitol (DTT). Free sulfhydryl groups in the polymers were then conjugated to maleimide-functionalized sHGP peptide in an acetononitrile/PBS solvent mixture (Scheme 1B).

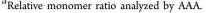
The control statistical copolymer of HPMA and $MaAhxK_{10}$, pHK₁₀, was used as a control material.

The sHGP-conjugated diblock copolymers (*b*-sHGP) contained about 5 sHGP per polymer based on the tryptophan

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Tab	le 1.	Po	lymer	Characterization	by	GPC	and AAA	I
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polymer	PDI	Mn (kDa, actual)	% sHGP ^a	% K ₁₀ ^a	% HPMA ^a			
pHPDSbHK ₁₀	1.2	97	-	16	84			
pHgsHGPbHK ₁₀ (block-sHGP)	-	-	4	10	86			
pHK ₁₀ sHGP (statistical-sHGP)	1.1	82	6	13	81			



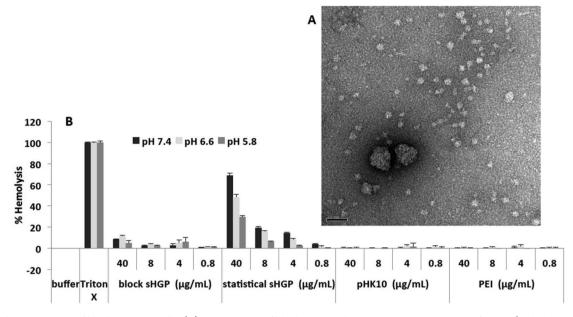
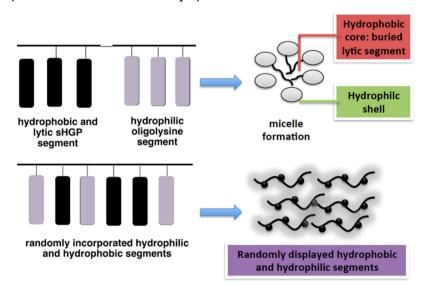


Figure 1. Characterization of block sHGP micelle. (A) TEM images of block sHGP solutions in water at 40K magnification (scale bar = 50 nm). (B) Hemolytic ability of polymers at various concentrations (μ g/mL).

Scheme 2. Proposed Polymer Architecture of sHGP Copolymers: Block vs Statistical



absorbance at 280 nm. All polymers were further characterized by gel permeation chromatography (GPC) and amino acid analysis (AAA) for polymer dispersity and peptide incorporation (Table 1). Interestingly, the diblock copolymers showed an early elution time suggesting the formation of a higher selfassembled structure (data not shown).

Equimolar solutions of sHGP-modified diblock and statistical copolymers were therefore characterized by transmission electron microscopy (TEM) and dynamic light scattering (DLS) analysis. No particles were detected by either method for statistical copolymer solutions. In contrast, TEM revealed nanoparticle structures in the diblock solutions (Figure 1A). The majority of particles was $\sim 10-20$ nm in diameter, although a small population of 50–100 nm particles were also observed. DLS analysis reported particles with mean hydrodynamic diameter ~ 250 nm, skewed by the population of larger particles (data not shown).

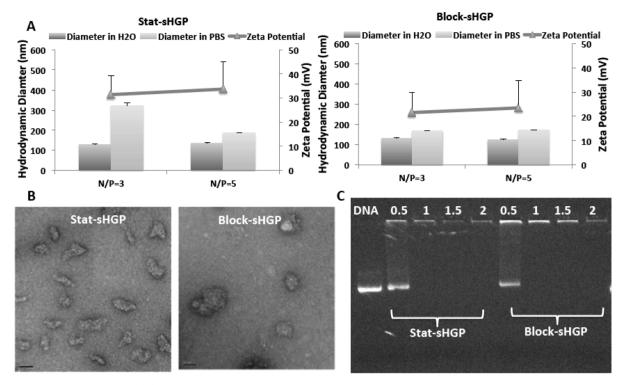


Figure 2. Polyplex characterization. (A) Sizing and zeta potential measurements of stat-sHGP and block-sHGP polyplexes at N/P ratios 3 and 5. (B) TEM images of stat-sHGP and block-sHGP polyplexes. TEM images were conducted on 400-mesh Formvar/copper grids stained with uranyl acetate. Scale bar represents 50 nm. (C) Agarose gel electrophoresis of polymer/DNA complexes prepared at different N/P ratios using stat-sHGP and block-sHGP. Lane 1 is free DNA; lanes 2–9 correspond to N/P ratios.

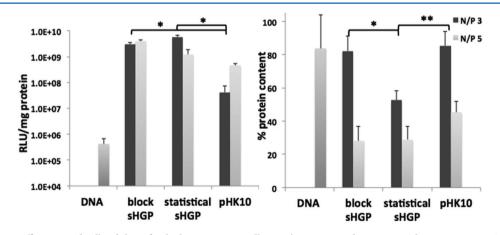


Figure 3. Transfection efficiency and cell viability of polyplexes to HeLa cells at N/P ratios 3 and 5. Data are shown as mean + SD (n = 4; Student's *t*-test, *p < 0.01, **p < 0.001).

The particle size reported by DLS for bimodal nanoparticles has been shown to be less accurate than imaging methods since the intensity of scattered light for particle with radius "r" is proportional to $r^{6.16}$ We evaluated the membrane lytic capability of sHGP copolymers, pHK₁₀, and PEI using an erythrocyte leakage assay (Figure 1B). The statistical sHGP showed a higher lytic ability at increasing polymer concentration (0.8–40 µg/mL) compared to the diblock analogue, while the two control polymers pHK₁₀ (a copolymer of HPMA and MaAhxK₁₀) and polyethylenimine (PEI) showed no lytic activity at the tested concentrations. These results led us to propose that the amphiphilic diblock copolymers that contain a hydrophobic sHGP block and a hydrophilic oligolysine block form micellar structures in aqueous solutions, thereby sequestering the lytic peptide (Scheme 2). Block copolymers, peptide, proteins, and lipids, containing hydrophobic and hydrophilic segments, are known monomers that can undergo self-assembly. In the controlled release field, self-assembled macromolecules have been utilized as constructs for drug delivery. Materials that undergo desolvation, collapse, and intermolecular hydrophobic interactions can potentially form self-assembled structures such as micelles, vesicles, lamellar sheets, and networks.^{17,18} Micelle constructs are commonly formed from block copolymers, wherein assembly of hydrophobic blocks results in a hydrophobic core shielded by the hydrophilic groups. Inverted micelles, with a hydrophilic outer shell and a polycationic inner core that interacts with the negatively charged nucleic acid, have been widely exploited in plasmid delivery.^{19–23} Polymeric or amphiphilic peptide-based micelles with a hydrophobic core shell

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have also been reported for improved gene delivery and better cell viability.^{24–26} Here, we evaluate a self-assembling micelle formed from an amphiphilic, peptide–polymer hybrid material for nucleic acid delivery.

The polymers were then complexed with plasmid DNA to form polyplexes with diameters ~200 nm in both water and physiologic salt concentrations (Figure 2A). These observations follow our previously reported data showing that HPMA-cooligolysines are salt stable at $DP \ge 100$.¹¹ Zeta potential measurements (Figure 2A) indicate that the net surface charge for both sHGP copolymers at N/P ratio 3 and 5 are both positive (20-30 ~kV). This is necessary for efficient polyplexmediated gene delivery in which the excess positive charge interacts with the negatively charged cellular membrane allowing for enhanced uptake and delivery.²⁷ TEM images at N/P ratio = 5 for both copolymers complexed with DNA showed short rods (st-sHGP) and spherical (b-sHGP) morphologies (Figure 2B). A gel retardation assay demonstrated that both copolymers completely associate with plasmid DNA at a similar polymer concentration (N/P ratio = 0.5)(Figure 2C).

The complexes were then used to transfect cultured mammalian cells. In vitro gene transfection showed that both the diblock and statistical copolymers show similar transfection efficiency that is significantly increased over the control pHK₁₀ copolymer; both block and statistical sHGP produced about 2 orders of magnitude higher reporter protein activity compared to pHK10 at N/P ratio = 3 and about an order of magnitude higher at a higher polymer concentration (N/P ratio = 5) (Figure 3). Thus, the incorporation of the endosomolytic agent, sHGP, into the copolymers containing HPMA and oligolysines further enhanced the gene delivery function of the prepared constructs, consistent with our previous report.⁸ Most notably, the diblock sHGP shows better cell viability compared to the statistical analogue.

There are a few possible explanations for the increased transfection efficiency observed with the diblock copolymer. First, the sHGP may be exposed intracellularly to enhance endosomal release. While the micelles did not show lytic behavior at acidic pH (Figure 1B), other endosomal conditions, such as competitive binding to intracellular proteins, may destabilize micelles and expose sHGP. Second, micellar morphologies may themselves increase delivery efficiency. For example, Yang and co-workers previously reported that amphiphilic constructs, made up of polyalanine as the hydrophobic block and polylysine and polyhistidine as the hydrophilic block, self-assemble into cationic and micellar nanoparticles in aqueous solutions. Condensation with DNA on the outer shell demonstrated stronger DNA-binding ability and resistance to enzymatic degradation compared to the control peptide without a hydrophobic block. Most importantly, increased cell viability and transfection efficiency were observed with these micellar structures.²⁴ The mechanism of increased delivery efficiency will be explored in future work.

In this communication, we have shown that the lytic activity of the truncated HGP further enhanced the gene delivery properties of the copolymers HPMA and oligolysines. Furthermore, we have demonstrated (for the first time) that peptide-based copolymers constructed in a diblock architecture, with the hydrophobic sHGP on one block and hydrophilic oligolysine on the other, are capable of self-assembly. This is an attractive feature for designing biocompatible and nontoxic nonviral vectors for nucleic acid delivery.

ASSOCIATED CONTENT

S Supporting Information

Experimental information including GPC, AAA, and hemolysis results is included. This material is available free of charge via the Internet at http://pubs.acs.org.

AUTHOR INFORMATION

Corresponding Author

*E-mail: spun@u.washington.edu.

Notes

The authors declare no competing financial interest.

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REFERENCES

(1) Varkouhi, A. K.; Scholte, M.; Storm, G.; Haisma, H. J. J. Controlled Release 2011, 151, 220.

- (2) Behr, J. P. Chimia 1997, 51, 34.
- (3) Wagner, E. Adv. Drug Delivery Rev. 1999, 38, 279.
- (4) Wolfert, M. A.; Seymour, L. W. Gene Ther. 1998, 5, 409.
- (5) Hoyer, J.; Neundorf, I. Acc. Chem. Res. 2012, 45, 1048.
- (6) Costin, J. M.; Rausch, J. M.; Garry, R. F.; Wimley, W. C. Virol J. 2007, 4, 123.
- (7) Kwon, E. J.; Bergen, J. M.; Pun, S. H. Bioconjugate Chem. 2008, 19, 920.
- (8) Kwon, E. J.; Liong, S.; Pun, S. H. Mol. Pharm. 2010, 7, 1260.
- (9) Merkel, O. M.; Urbanics, R.; Bedocs, P.; Rozsnyay, Z.; Rosivall, L.; Toth, M.; Kissel, T.; Szebeni, J. *Biomaterials* **2011**, *32*, 4936.

(10) Ogris, M.; Brunner, S.; Schuller, S.; Kircheis, R.; Wagner, E. Gene Ther. **1999**, 6, 595.

(11) Johnson, R. N.; Chu, D. S.; Shi, J.; Schellinger, J. G.; Carlson, P. M.; Pun, S. H. J. Controlled Release 2011, 155, 303.

(12) Schellinger, J. G.; Pahang, J. A.; Johnson, R. N.; Chu, D. S.; Sellers, D. L.; Maris, D. O.; Convertine, A. J.; Stayton, P. S.; Horner, P. J.; Pun, S. H. *Biomaterials* **2013**, *34*, 2318.

(13) Shi, J.; Schellinger, J. G.; Johnson, R. N.; Choi, J. L.; Chou, B.; Anghel, E. L.; Pun, S. H. *Biomacromolecules* **2013**, *14*, 1961.

- (14) Chu, D. S.; Schellinger, J. G.; Shi, J.; Convertine, A. J.; Stayton,
- P. S.; Pun, S. H. Acc. Chem. Res. 2012, 45, 1089.
- (15) Ahmed, M.; Narain, R. Prog. Polym. Sci. 2013, 38, 767.
- (16) Hoo, C. M.; Starostin, N.; West, P.; Mecartney, M. L. J. Nanopart. Res. 2008, 10, 89.
- (17) Branco, M. C.; Schneider, J. P. Acta Biomater. 2009, 5, 817.
- (18) Chen, S.; Cheng, S. X.; Zhuo, R. X. Macromol. Biosci. 2011, 11, 576.
- (19) Harada-Shiba, M.; Yamauchi, K.; Harada, A.; Takamisawa, I.; Shimokado, K.; Kataoka, K. *Gene Ther.* **2002**, *9*, 407.

(20) Miyata, K.; Fukushima, S.; Nishiyama, N.; Yamasaki, Y.; Kataoka, K. J. Controlled Release **2007**, *122*, 252.

(21) Wei, H.; Volpatti, L. R.; Sellers, D. L.; Maris, D. O.; Andrews, I. W.; Hemphill, A. S.; Chan, L. W.; Chu, D. S.; Horner, P. J.; Pun, S. H. *Angew. Chem., Int. Ed. Engl.* **2013**, *52*, 5377.

(22) Wei, H.; Schellinger, J. G.; Chu, D. S.; Pun, S. H. J. Am. Chem. Soc. 2012, 134, 16554.

(23) Convertine, A. J.; Diab, C.; Prieve, M.; Paschal, A.; Hoffman, A. S.; Johnson, P. H.; Stayton, P. S. *Biomacromolecules* **2010**, *11*, 2904.

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(24) Wiradharma, N.; Khan, M.; Tong, Y. W.; Wang, S.; Yang, Y. Y.

(24) Wiradharma, N.; Khan, M.; Tong, Y. W.; Wang, S.; Yang, Y. Y. Adv. Funct. Mater. 2008, 18, 943.
(25) Guo, X. D.; Tandiono, F.; Wiradharma, N.; Khor, D.; Tan, C. G.; Khan, M.; Qian, Y.; Yang, Y. Y. Biomaterials 2008, 29, 4838.
(26) Seow, W. Y.; Yang, Y. Y. J. Controlled Release 2009, 139, 40.
(27) Richard, J. P.; Melikov, K.; Vives, E.; Ramos, C.; Verbeure, B.; Gait, M. J.; Chernomordik, L. V.; Lebleu, B. J. Biol. Chem. 2003, 278, 505 585.